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Reaction of 2,4-(Naphthalene-1,8-Diyl)-1,3,2,4-Dithiadiphosphetane 2,4-with Bases and Acids

ZHAOFU FEI, ALEXANDRA M. Z. SLAWIN and J. DEREK WOOLLINS

Department of Chemistry, St. Andrews University, Fife, Scotland KY16 9ST, UK

The compound 2,4-(naphthalene-1,8-diyl)-1,3,2,4-dithiadiphosphetane 2,4-disulfide 1 reacts with different bases 2a-2d HNR(HNR = N,N,N',N'-tetramethyl-guanidine, morpholine, pyrrolidine or NH₃) to give the corresponding ammonium salts of 3a-3d. The reaction of 1 with Et₃N-3HF lead to the formation of 4a. The new compounds formed from the reactions have been studied spectroscopically and by X-ray crystallography.

Keywords: P-S bond; NMR; X-Ray crystal structure

Introduction

Organophosphorus-sulfur heterocycles with the P_2S_2 systems have attracted great interest in recent studies^{1,2}. In continuation of this research³, we studied the propertities of 2,4-(naphthalene-1,8-diyl)-1,3,2,4-dithiadiphosphetane 2,4-disulfide 1 towards different bases and acids (HF, HCl, HBr) in organic solvents.

Results and Discussion

The reaction of 1 with 2a (N,N,N',N'-tetramethylguanidine) in THF or dichloromethane gave compound 3a (Fig.1) in nearly stochiometric yield. The second P-S-P bond could not be broken even excess of guanidine was used.

Figure 1: X-Ray crystal structure of 3a

The analogous reactions of 1 with 2b (morpholine), 2c (pyrrolidine) and 2d (NH₃(g)) in a solution of dichloromethane gave the corresponding compounds 3b, 3c and 3d in nearly quantitative yield (Scheme 1).

Scheme 1: Compounds 3b, 3c and 3d

Compounds 3b-3d were initially soluble in the dichloromethane but the solubility decreased after removal of the solvent. In the ³¹P{H} NMR the ²J(PP) are in the range 11.24-13.12 Hz. All 3a-3d are very stable in inert atmosphere, but they seem to undergo slow decomposition in dimethyl sulfoxide.

Treatment of 1 with morpholine in the presence of excess triethyl amine lead to the formation of a mixture of several compounds, from which 3e (Fig. 2) could be isolated by recrystallisation. The rest of the material which was poorly soluble in common solvents like diethyl ether, dichloromethane, chloroform, acetonitrile and dimethyl sulfoxide and could not be completely identified.

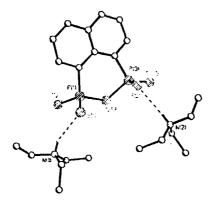


Figure 2: X-Ray crystal structure of 3e

If 1 was treated with Et₃N·3HF, 4 (Fig.3) could be obtained in high yield; however, the treatment of 1 with HCl and HBr under the same condition did not lead to the similar results.

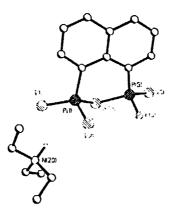


Figure 3: X-Ray crystal structure of 4

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